

Aflatoxin in food and feed: occurrence, legislation and inactivation by physical methods

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Aflatoxins are secondary metabolites of the molds *Aspergillus flavus* and *Aspergillus parasiticus.* They are highly toxic, mutagenic, teratogenic and carcinogenic compounds found to contaminate a wide variety of important agricultural products such as peanuts, maize, rice and cottonseed. Aflatoxins are found in many countries, especially in tropical and subtropical regions where conditions of temperature and humidity are optimum for growth of the molds and for production of the toxin. Removal or inactivation of aflatoxin in food and feedstuffs is a major global concern. Aflatoxins can be detoxified or removed from contaminated foods and feeds by physical, chemical and biological methods. This article reviews some aspects of aflatoxins concerning their occurrence in food and feedstuffs, legislation and detoxification by physical methods. The physical methods include solvent extraction, adsorption, heat treatment and irradiation. 0 1997 Elsevier Science Ltd. All rights reserved

INTRODUCTION

The problem of food and feed contamination with aflatoxins is of current concern and has received a great deal of attention during the last three decades. The frequent incidence of these toxins in agricultural commodities has a potential negative impact on the economies of the affected regions, especially in the developing countries where harvest and post-harvest techniques adequate for the prevention of mold growth are seldom practised.

Aflatoxins are highly toxic, mutagenic, teratogenic and carcinogenic compounds that have been implicated as causative agents in human hepatic and extrahepatic carcinogenesis (Massey et *al.,* 1995). There is enough evidence from epidemiological studies regarding the hazardous nature of aflatoxins, as potent carcinogens, in connection with incidence of liver cancer in populations exposed to the toxin from contaminated foods (Shank *et al.,* 1972; Peers & Linsell, 1973; Dichter, 1984; Groopman *et al.,* 1988). Human exposure to aflatoxins can result directly from ingestion of contaminated foods, or indirectly from consumption of foods from animals previously exposed to aflatoxins in feeds. Moreover, aflatoxins have been associated with incidence of the disease kwashiorkor, a consequence of protein energy malnutrition, in children (Adhikari et *al.,* 1994).

Ingestion of aflatoxins leads to substantial loss of productivity and degradation of meat quality in farm animals consuming contaminated feeds (Bonomi et *al.,* 1993, 1994). In 1986 there was an estimated loss of about 140 million US dollars as a direct sequence of weight loss in broilers consuming low levels of mycotoxins (Palmgren & Hayes, 1987).

Aflatoxins are secondary metabolites produced by species of *Aspergillus,* specifically *Aspergillus flavus* and *Aspergillus parasiticus,* which are found worldwide in air and soil and are found to infest both living and dead plants and animals. The discovery of aflatoxins dates back to the year 1960 following the severe outbreak of the turkey 'X' disease, in the UK, which resulted in the deaths of more than 100000 turkeys and other farm animals. The cause of the disease was attributed to a feed, containing Brazilian peanuts, which was heavily infested with *A. javus.* Subsequent analysis of the feed using thin-layer chromatography (TLC) revealed that a series of fluorescent compounds, later termed aflatoxins, were responsible for the outbreak (Sargeant *et al.,* 1961; Bash & Rae, 1969; Davis & Diener, 1979). Hence, the name aflatoxin, an acronym, has been formed from the following combination: the first letter, 'A' for the genus *Aspergillus,* the next set of three letters, 'FLA', for the species $flavus$, and the noun 'TOXIN' meaning poison (Ellis et *al.,* 1991).

Efforts have been made to remove or detoxify aflatoxins in contaminated foods and feedstuff (Doyle *et al.,* 1982; Samarajeewa *et al.,* 1990; Beaver, 1991). This paper reviews some aspects of aflatoxins concerning their occurrence in food and feedstuffs, legislation and detoxification by physical methods.

OCCURRENCE

Many agricultural commodities in general, and peanuts in particular, are liable to infestation by aflatoxigenic molds, and thereby contamination with aflatoxins. In this context, aflatoxicosis is becoming a worldwide problem. Other important crops such as maize, rice and cottonseed have also been found to be infested by aflatoxigenic molds in a number of different countries (Table 1). The incidence of aflatoxins in foods and feeds is relatively high in tropical and subtropical regions (Table 2), where the warm and humid weather provides optimal conditions for the growth of the molds. The growth of *A. flaws* and production of aflatoxin in natural substrates are influenced by a number of factors, including the type of substrate, fungal species, moisture content of the substrate, presence of minerals, relative humidity of the surroundings, temperature and physical damage of kernels (Viquez *et al.,* 1994). Production of aflatoxin B_1 (AFB1) by *A. flavus* and *A. parasiticus* is higher in rice than in peanut (Pinto *et al.,* 1991). The minimum temperature range for *A. parasiticus* growth is $6-8^{\circ}$ C, the maximum is $44-46^{\circ}$ C, and the optimum is 25-35°C (Diener *et al.,* 1982). *A. flaws* can produce aflatoxin at temperatures of $12-42^{\circ}$ C, but the optimum is $28-30$ °C (Brackett, 1989). The minimum moisture content of peanuts necessary for *A. fluvus* growth is 8-

10% at about 82% relative humidity, and aflatoxin production on peanuts is optimum at $15-35%$ moisture content (Brackett, 1989). Immature, broken, undersized, loose-shelled, rancid and discolored peanut kernels are most likely to be contaminated, in comparison with mature sound kernels (Davis & Diener, 1979; Beaver, 1991; Chiou *et al.,* 1994; Rucker *et al.,* 1994). Physical damage of peanut shell facilitates its penetration by the molds and thereby contamination with aflatoxin (Porter *et al.,* 1986). The physical damage can also improve the hygroscopic characteristics of peanuts during storage, thus leading to higher incidence of the molds as a result of increased moisture content of the kernels (Chiou *et al.,* 1984). Exposure of peanut pods to drought stress also increases the risk of contamination (Sanders *et al.,* 1993).

Peanuts of all varieties in all geographical locations are liable to contamination with aflatoxins in the field before and after harvest, during curing and storage. Some documented cases of aflatoxin incidence in peanuts and peanut products are listed in Table 3. The reason why the incidence is more frequent in peanuts than in other agricultural commodities is not fully understood. However, it could be attributed to the fact that *A. flavus* dominates the mycoflora of peanut field soils, where the kernels develop and mature beneath the surface (Diener *et al.,* 1982).

INACTIVATION

With increasing knowledge and awareness of aflatoxin as a potent source of health hazards to both man and farm animals, a great deal of effort has been made to completely eliminate the toxin or reduce its content in foods and feedstuffs to significantly lower levels.

Commodity	Country	Species	Reference
Peanuts	Sudan	A. flavus	Elamin et al., 1988
	Egypt	$A.$ flavus + $A.$ niger	Moubasher et al., 1980
	S. Africa	A. flavus + A. parasiticus	Dutton & Westlake, 1985
Maize	India	A. flavus	Gaur & Siradhana, 1989
	China	A. flavus	Zhen-Zhen, 1989
	Uganda	$A.$ flavus + $A.$ parasiticus	Sebunya & Yourtee, 1990
	Nigeria	A. flavus + A. parasiticus + A. niger	Aja Nwachukwu & Emejuaiwe, 1994
	USA	A. flavus	Guo et al., 1995
Wheat	China	A. flavus	Zhen-Zhen, 1989
	Russia	A. flavus	L'Vova et al., 1993
Rice	China	A. flavus	Zhen-Zhen, 1989
	India	$A.$ flavus + $A.$ parasiticus	Jayaraman & Kalyanasundaram, 1990
Millet	India	A. flavus + A. parasiticus	Mishra & Daradhiyar, 1991
Soybean	Argentina	$A.$ flavus + $A.$ parasiticus	Pinto et al., 1991
Sunflower oil	China	A. flavus	Zhen-Zhen, 1989
	Russia	A. flavus	L'Vova et al., 1993
Coconut	India	A. flavus	Bilgrami & Choudhary, 1993
Pistachio nuts	USA	A. niger + A. flavus + A. parasiticus	Doster & Michailides, 1994
	Turkey	A. flavus	Heperkan et al., 1994
Figs	Switzerland	A. flavus + A. parasiticus	Steiner et al., 1988
Mustard seed	India	A. flavus	Sahay & Prasad, 1990

Table 1. Occurrence of *Aspergillus* **in some agricultural commodities**

duction of good crop husbandry and appropriate cul-
tural practices that limit the growth of aflatoxigenic fungi. Post-harvest contamination can be minimized by application of proper curing, drying, sorting and

Pre-harvest contamination can be reduced by intro-

storage procedures. However, the contamination is

intition of good crop husbandry and appropriate cul-

often unavoidable and still remains a serious problem associated with many important agricultural commodities, which emphasizes the need for a suitable process to inactivate the toxin. In addition to the capability of a

 $u -$, not mentioned.

 $h \overrightarrow{B1} + G1 + B2 + G2.$

Table 3. Occurrence of aflatoxin in peanuts and peanut products

 $\overset{a}{\ }$, not mentione

 b B₁ + G₁ + B₂

process to degrade the toxin to safe levels, it should meet the following requirements (Jemmali, 1989; Ellis *et al..* 1991; Park & Liang, 1993):

- It must not result in the formation of other toxic substances or leave any harmful residues that might diminish the overall safety of the treated product.
- The nutritional quality of the product should not be seriously suppressed.
- It should not adversely affect desirable physical and sensory properties of the product.
- It has to be economically feasible, and technically applicable.
- It must be capable of destroying the spores and mycelia of aflatoxigenic fungi, if they are present in the product, which might, under favourable conditions, proliferate and reproduce the toxin.

A number of methods have been investigated in connection with their ability to inactivate aflatoxins in contaminated food and feedstuffs. These methods aim at either removing the toxin from the food or destroying it in the food. They can be classified into chemical, biological and physical methods.

A large number of chemicals can react with aflatoxins and convert them to less toxic and mutagenic compounds. These chemicals include acids, bases, oxidizing agents, bisulphites and gases. However, most of the chemical processes that have been investigated are impractical (carried out under drastic conditions of temperature and pressure), unsafe (form toxic residues) and unfavourable (degrade the nutritional, sensory and functional properties of the product). Currently, ammoniation and treatment with sodium bisulphite are the major industrial processes widely used to inactivate peanut meal, maize and cottonseed destined for animal feeding.

Many microorganisms including bacteria and acidproducing molds can metabolize and inactivate aflatoxins, with *Flavobacterium aurantiacum* as the most active organism. It was postulated that the inactivation was a result of acid production and subsquent conversion of AFBl to AFB2a, which is 1000 times less mutagenic than the parent toxin (Park & Liang, 1993).

Inactivation by physical methods involves extraction with solvents, adsorption, inactivation by heat and irradiation. The physical methods are discussed in detail in the following sections.

Extraction

Extraction with solvents has been used to remove aflatoxins from the oilseeds peanut and cottonseed (Dollear, 1969; Hron *et al.,* 1992, 1994). Materials treated in this way may only be suitable for animal feeding. The solvents used include 95% ethanol, 90% aqueous acetone, 80% isopropanol, hexane-methanol, methanol-water, acetonitrile-water, hexane-ethanol-water

and acetone-hexane-water. The solvent:sample ratio was found to be crucial for recovery of the toxin (Cole & Dorner, 1994). Solvent extraction can remove all traces of aflatoxin from oilseed meals with no formation of toxic byproducts or reduction in protein content and quality. However, large-scale application of this technique is limited by high cost and problems related to disposal of the toxic extracts.

Adsorption

Some adsorbents can bind and thus remove aflatoxins from aqueous solutions. Bentonite clay adsorbed AFBl from a Sörensen buffer solution (30°C, 5 days). Separation of the clay resulted in the removal of $94-100\%$ of the toxin from the solution. Particle size reduction and pre-heating of the clay increased its ability to adsorb and retain the toxin (Masimango *et al.,* 1978). Also, bentonite removed 65-79% of AFMl from milk (Doyle *et al.,* 1982). Hydrated sodium calcium aluminosilicate (HSCAS; NovaSil clay) was reported (Phillips *et al.,* 1988) to have a high affinity for AFBl. The HSCAS removed more than 80% of the toxin from solution. In *vivo* studies demonstrated the role of HSCAS in preventing the mutagenicity and toxicity of AFBl (Phillips *et al.,* 1988).

Heat

Aflatoxins have high decomposition temperatures ranging from 237°C to 306°C. Solid AFBl is quite stable to dry heating at temperatures below its thermal decomposition temperature of 267°C (Betina, 1989). The use of heat to inactivate aflatoxin in contaminated food has been attempted. Normal home cooking conditions such as boiling and frying (approx. 150° C) failed to destroy AFBl and AFGl in the solid state (Kamimura, 1989). Table 4 summarizes the results of some studies conducted in connection with the degradation of aflatoxin in foods by different heat treatments. Temperatures above 150°C were necessary to attain partial destruction of the toxin. The extent of the destruction achieved was very dependent on the initial level of contamination, heating temperature and time. Moreover, the type of food and aflatoxin also influenced the degree of inactivation achieved. Degradation of aflatoxin by heat is also governed by the moisture content, pH and ionic strength of the food. The moisture content is a critical factor; contaminated foods that contain more moisture can more easily be inactivated by heat. Mann *et al.* (1967) observed that heating a cottonseed meal containing 30% moisture at 100°C for 1 h degraded 74.8% of aflatoxins $(B1 + B2)$ present in the meal, whereas only 32.7% of the toxins were destroyed after heating a similar meal containing 6.6% moisture, under the same conditions. The sensitivity of AFBl to moistheat was also demonstrated in irradiated foods. Nkama and Muller (1988) studied the effect of the moisture content of a rice meal on destruction of AFBl in the meal after exposure to a tungsten lamp light (intensity 43 mW cm⁻²) at 36°C for 2 h. They observed that increasing the moisture content of the meal from 14.1% to 18.7% increased the rate of AFBl destruction from 40% to 63%. It has been suggested that the presence of water helps in opening the lactone ring in AFBl (by the addition of a water molecule to the ring) to form a terminal carboxylic acid. The terminal acid group thereafter undergoes heat-induced decarboxylation (Coomes *et al.,* 1966). The presence of ionic salts increased the extent of aflatoxin degradation by heat. Farah et al. (1983) cooked raw unshelled peanuts in a 5% NaCl solution in an autoclave at 116"C, 0.7 bar for 30 min. This treatment reduced the total content of aflatoxins $(B1 + G1 + B2 + G2)$ by 80-100%. The removal of the toxins was attributed to the addition of NaCl, as compared to unsalted controls.

AFM1 is a monohydroxy derivative of AFB1 secreted in milk by the cow after ingestion of feeds containing

AFB1. The effects of pasteurization and sterilization on reduction of AFMl in milk were studied (Table 5). The percentage reduction in AFMl levels reported by different researchers are conflicting. For example, pasteurization of milk at 62°C for 30 min was observed (Purchase *et al.,* 1972) to reduce the AFMl content in milk by 32%, whereas a similar treatment did not reduce the toxin content, as concluded by Stoloff *et al.* (1975). The contradiction could be attributed to the following:

AFMl is not stable in milk; its concentration spontaneously declines with time. About 40% of AFMI in naturally contaminated milk disappeared after 4 days and about 80% after 6 days storage at 0°C (McKinney *et al.,* 1973). Kiermeier and Mashaley (1977) reported a reduction ranging from 18.8% to 24.2% in AFMl content in milk upon storage at 5°C for l-3 days. Also, a 34% decrease in AFMl content in milk over 2 weeks

 α N, natural; A, artificial.

 b B₁ + G₁ + B₂ + G₂.

 α A, artificial; N, natural.

storage at 21°C was reported (Heimbecher et *al.,* 1988). The poor stability of AFMl in milk probably leads to contradictory results when the concentration of AFMl in milk is assayed by different individuals at different laboratories.

- Differences in the initial level of contamination, and in the methods used to extract, clean-up and assay the toxin. Not all the methods are of equal accuracy, precision and sensitivity (Mehan *et al.,* 1985).
- Status of contamination, i.e. whether the treated milk was naturally or artificially contaminated. It is easier to inactivate aflatoxins in artificially contaminated substrates than in those naturally contaminated.

Rustom *et al.* (1993) studied the effects of pH (5.0, 8.0, 10.2), temperature (121°C, 130°C, 140°C) and heating time (5 s, 20 s, 15 min) on mutagenic activity (assayed by Ames test) of peanut beverages artificially contaminated with AFBl. Heat treatments at pH 8.0 were not effective in reducing the mutagenic activity. On the other hand, the treatments pH 10.2, 130 \degree C, 20 s and pH 10.2, 121° C, 15 min reduced the mutagenic activity by 78% and 88%, respectively. The inactivation was attributed to partial conversion of AFBl to AFDl through hydrolysis of the lactone ring in AFBl, catalysed by NaOH used to adjust the pH. AFDl is 450 times less mutagenic than AFBl. The treatments pH 5.0, 130 $^{\circ}$ C, 20 s and pH 5.0, 121 $^{\circ}$ C, 15 min reduced the mutagenic activity by 76% and 73%, respectively. The inactivation was attributed to partial hydration of AFBI, at the terminal furan ring, to AFB2a catalysed by HCl used to adjust the pH. AFB2a is 1000 times less mutagenic than AFBl. Changing the pH, from 8.0 to 5.0 or 10.2, without heating did not significantly reduce the mutagenic activity, thus emphasizing the synergistic role of pH and heat in reducing the mutagenic activity.

Microwave heating showed great potential for destruction of aflatoxin in contaminated peanuts depending on power level and treatment time. For instance, roasting an artificially contaminated peanut meal in a microwave at 6 kW for 4 min destroyed 95% of aflatoxin in the meal (Staron et *al.,* 1980). Similarly, microwave heating also reduced the level of aflatoxin in peanut kernels (Luter *et al.,* 1982). At lower energy $(0.7 \text{ kW}, 8.5 \text{ min})$, $48-61\%$ of AFB1 was destroyed in artificially contaminated peanut kernels. However, the same treatment resulted in only 30-45% reduction of AFBl in naturally contaminated kernels (Pluyer *et al.,* 1987).

Irradiation

Radiation is classified into two categories: ionizing and non-ionizing. In ionizing radiation (e.g. X-rays, gamma rays, ultraviolet rays) potential changes may occur in molecules of the irradiated object with little or virtually no temperature rise. These molecular changes might be quite harmful to living organisms exposed to large doses of ionizing radiation. On the other hand, non-ionizing radiation (e.g. radio waves, microwaves, infrared waves, visible light) in sufficient intensity leads to a rise in temperature, usually accompanied by molecular changes that are of no hazardous nature to man. The use of ionizing radiation to free foods from pathogenic microorganisms is among the methods applicable in food preservation (Kyzlink, 1990). Despite the debate on safety of irradiated foods in connection with human health, however, food irradiation is becoming a technique of potential application on a commercial scale to render food products sterile (Diehl, 1990).

Ultraviolet (UV) light

Aflatoxins are sensitive to UV radiation. AFBl absorbs UV light at 222, 265 and 362 nm, with maximum absorption occurring at 362 nm, which may lead to the formation of up to 12 photodegradation products (Samarajeewa *et al.,* 1990). AFBl and AFGl underwent photochemically driven series of reactions when exposed to UV light (365 nm, 1 h) on silica gel TLC plates. The photodegradation products were less toxic, to chick embryos, than the parent toxins (Andrellos *et al.,* 1967). Treatment of peanut oil with UV light for 2 h destroyed $40-45%$ of aflatoxins initially present in the oil (Shantha & Murthy, 1977). Exposure of artificially contaminated milk to UV light inactivated $3.6-100\%$ of AFM1 in the milk, depending on exposure time $(2 -$ 60 min). Also, addition of hydrogen peroxide (1%) to the UV-irradiated milk (10 min) completely (100%) destroyed AFMl (Yousef & Marth, 1985). Destruction of AFM 1, in aqueous solution, by UV energy was of first-order irreversible reaction kinetics, and it was attributed to opening of the double bond in the terminal furan ring in AFMl (Yousef & Marth, 1987). Also, UV irradiation (30 min) of dried figs artificially contaminated with AFB1 (250 μ g kg⁻¹) reduced the toxin level by 45.7% (Altug *et al.,* 1990).

Gamma rays

The use of gamma radiation to inactivate aflatoxins was investigated. The toxicity of a peanut meal contaminated with AFBl was reduced by 75% and 100% after irradiation with gamma rays at a dose of 1 and 10 kGy, respectively (Temcharoen & Thilly, 1982). However, doses higher than 10 kGy inhibited the seed germination, and increased the peroxide value of the oil in gamma-irradiated peanuts (Chiou *et al.,* 1990). The presence of water has an important role in the destruction of aflatoxin by gamma energy, since radiolysis of water leads to the formation of highly reactive free radicals. These radicals can readily attack AFB1, at the terminal furan ring, giving products of lower biological activity. The mutagenic activity of AFBl in an aqueous

solution (5 μ g ml⁻¹ water) was reduced by 34%, 44%, 74% and 100% after exposure to gamma rays at 2.5, 5, 10 and 20 kGy, respectively (Van Dyck *et al.,* 1982). Also, a dose of 10 kGy completely (100%) inactivated AFBl, and destroyed 95% of AFGl in a dimethylsulphoxide-water (1:9, v/v) solution (Mutluer & Erkoc, 1987). Addition of 1 ml of 5% hydrogen peroxide to an aqueous AFB1 solution (50 μ g ml⁻¹) resulted in 37-100% degradation of the toxin at a lower dose (2 kGy). The final degradation products showed no biological activity in Ames mutagenicity test. The same treatment reduced the level of AFBl in peanut kernels by 73-80% (Pate1 *et al., 1989).*

Solar irradiation

Solar energy also destroyed aflatoxins in some foods (Table 6). Studies conducted in India revealed the efficacy of sunlight in degrading the toxin in different peanut products. It seems that UV rays from sunlight play an important role in the photodestruction of aflatoxin. Peanut protein can bind aflatoxins (Shantha & Murthy, 1980, 1981), and aflatoxins bound to the protein appear to be less susceptible to photodegradation than the free toxin. About 90% of AFBl in artificially contaminated peanut flakes was destroyed by sunlight, whereas only 50% of the toxin was destroyed by the same treatment when it was present as a natural contaminant (Shantha *et al., 1986).* Probably, aflatoxins were more bound to proteins in the naturally contaminated flakes than in the artificially contaminated flakes, which made them less liable to photodegradation by sunlight.

INTERNATIONAL LEGISLATION

The hazardous nature of aflatoxin to humans and animals has necessitated the need for establishment of control measures and tolerance levels by national and international authorities. Different countries have different regulations for aflatoxin. The general trend is that industrialized countries usually set lower tolerance levels than the developing countries, where most of the susceptible commodities are produced. For example, the

tolerance level for aflatoxin in foods is $5 \mu g kg^{-1}$ in Sweden (Akerstrand & Möller, 1989) and 10 μ g kg⁻¹ in Japan (Aibara & Maeda, 1989), whereas it is 30 pug kg-' in Brazil (Sabino *et al., 1989a).* However, such lack of harmony may give rise to difficulties in the trade of some commodities.

The first legislative act was undertaken in 1965 by the Food and Drug Administration (FDA) of the USA, which proposed a tolerance level of 30 μ g kg⁻¹ of total aflatoxins $(B1 + G1 + B2 + G2)$. With increasing awareness of aflatoxins as potent toxic substances, the proposed level was lowered to 20 μ g kg⁻¹ in 1969 (FDA, 1977). Since then, the tolerance levels have been revised and re-evaluated occasionally. The current tolerance levels established by the FDA for food and feeds are listed in Table 7.

In 1973, the European Economic Community (EEC) established legislation on maximum permitted levels of AFBl in different types of feedstuffs (EEC, 1974). The legislation has been frequently amended since then. However, the main framework of the legislation remains more or less the same. Table 8 lists the maximum allowable levels of AFBl in animal feeds as established by the EEC. The legislation has been in action since 30 November 1991 (EEC, 1991).

Contamination of agricultural commodities with aflatoxin, and hence establishment of regulations, has been a major concern to the Joint FAO/WHO Expert Committee. The Committee recommends that the presence of aflatoxin in food should be limited to 'irreducible levels'. An irreducible level is defined as: 'the concentration of a substance that can not be eliminated from a food without involving the discarding of that food altogether, or severely compromising the ultimate availability of major food supplies' (FAO/WHO, 1987). The current regulations for aflatoxin established by the Joint FAO/WHO Committee is given in Table 9.

The Dutch authorities undertook a worldwide enquiry, during 1986 and 1987, in which 66 countries were requested to report on their regulations and tolerance limits for mycotoxins in food and feeds. Of the 66 countries, 50 had enacted or proposed legislation for aflatoxins in foods, 35 for aflatoxins in feedstuffs, and 14 for aflatoxin Ml in milk and dairy products. An

Exposure time	Country	Food	Destruction $(\%)$	Reference
6 h	India	Peanut cake ^a	50	Shantha & Murthy, 1981
6 h	India	Case in ^a	83	Shantha & Murthy, 1981
10 _h	Nigeria	Maize, millet ^b	30.16	Okonkwo & Nwokolo, 1978
14 _h	India	Peanut flakes ^a	90	Shantha et al., 1986
14 h	India	Peanut flakes ^b	50	Shantha et al., 1986
15 min	India	Peanut oil ^b	99	Shantha & Murthy, 1977
10.40 min	USA	Olive oil ^a	55, 95	Mahjoub & Bullerman, 1988

Table 6. Destruction of aflatoxins in foods by solar energy

^a Artificially contaminated.

 b Naturally contaminated.</sup>

Table 7. FDA tolerance levels for total aflatoxin **CONCLUSIONS**

Item	Tolerance level $(\mu g \; kg^{-1})$
Food for human consumption	20
Feed for beef cattle and poultry	300
Feed for swine	200
Feed for breeding livestock	100
Feed for dairy cattle	20
Milk	በ ና

Source: Park & Njapau (1989); Park & Liang (1993).

Table 8. Maximum permitted level of aflatoxin Bl in different animal feeds established by the EEC"

Feed	Level $(\mu g \text{ kg}^{-1})^b$
Straight feeds ^c	
Peanut, copra, palm kernels, cotton	20
seed, babassu, maize and products	
derived therefrom	
Compound feeds ^d	
Complete feeds	
Feeds for cattle, sheep and goat	50
(except dairy cattle, calves and lambs)	
Feeds for pigs and poultry	20
(except young animals)	
Other complete feeds	10
Complementary feeds	
Feeds for cattle, sheep and goats	50
(with exception of dairy animals)	
Feeds for pigs and poultry	30
Other complementary feeds	5
(with exception of young animals)	

^a EEC (1991).

b Based on feed moisture content of 12%.

' Feed composed of a single item.

d Feed composed of a mixture of items.

Table 9. Aflatoxin legislations set by the Joint FAO/WHO Expert Committee"

Aflatoxin	Tolerance level $(\mu g \; kg^{-1})$	Food/feed
B1		Feed for dairy cattle
M ₁	0.05	Milk
$B1 + G1 + B2 + G2$	15	Raw peanut for
$B1 + G1 + B2 + G2$	10	human consumption Processed peanut for human consumption

"FAO/WHO (1990, 1992).

extensive overview of aflatoxin regulations in these countries is provided by Van Egmond (1989), later updated by Stoloff et al. (1991). The study revealed that, in most of the countries surveyed, the rationales for establishment of tolerance limits and regulations for aflatoxins were based on vague unsupported statements of carcinogenic risk for humans due to exposure to aflatoxins.

No doubt consideration of preventive measures aimed at reducing infestation of agricultural commodities with aflatoxgenic molds is the best way to control contamination with aflatoxin. In addition to application of proper cultural, storage and segregation practices, the development of crop species with high resistance to aflatoxigenic molds is of great potential in this context. Establishment of a unique detoxification method for all foods and feedstuffs in not possible. The efficiency of a method to inactivate aflatoxins depends, to a great extent, on the nature of the food, its moisture content, type of aflatoxin, level of contamination and degree of association of aflatoxins with the food constituents.

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